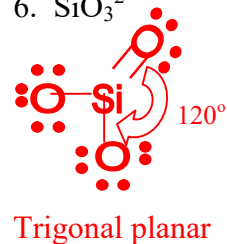
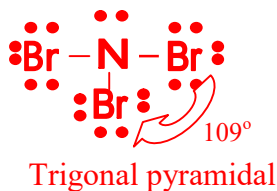
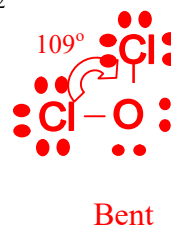
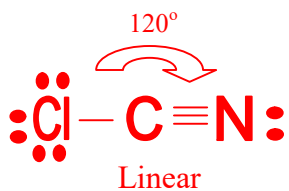
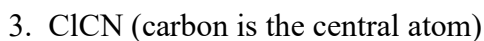
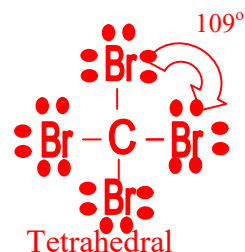


Shape Practice

Name: _____

Date: _____

Draw a structure for each of the following. On each structure label the approximate bond angle. Also name the geometry (i.e. “tetrahedral”, “bent”, etc.).



7. A certain molecule is bent and has a bond angle of about 109° . Is the molecule SeS_2 or SeCl_2 ?

The molecule is SeCl_2 because SeS_2 has bond angles of 120° .